Microwave-assisted Synthesis of Hybrid Heterocycles as Potential Anticancer Agents

Avula Srinivas,* Malladi Sunitha, Kammachichu Raju, Banothu Ravinder, Siluveru Anusha, Thallapalli Rajasri, Pothuganti Swapna, Dupa Sushmitha, Deva Swaroopa, Gurala Nikitha and Chakunta Govind Rao

Department of Chemistry, Vaagdevi Degree & PG College
Kishanpura, Warangal, Telangana, India 506001

* Corresponding author: E-mail: avula.sathwikreddy@gmail.com

Received: 24-12-2016

Abstract

In a one pot procedure, a series of novel hybrid heterocycles 6a–g and 7a–g were prepared by condensation of (3aS,4S,6S,6aS)-6-((1-(4-chlorophenyl)-1H-1,2,3-triazol-4-yl)methoxy)-2,2-dimethyltetrahydrofuro[3,4-d][1,3]dioxole-4-carbaldehyde 5 with mercapto acids and primary amines in the presence of ZnCl₂ under both microwave irradiation and conventional heating conditions. Compound 5 was prepared from di-acetone D-mannose via a click reaction, primary acetonide deprotection and oxidative cleavage. Characterization of new compounds has been done by IR, NMR, MS and elemental analysis. Anticancer activity of the compounds has also been evaluated.

Keywords: D-mannose, click reaction, cyclisation, anticancer activity

1. Introduction

1,2,3-Triazoles are one of the most important classes of heterocyclic organic compounds, which are reported to be present in a plethora of biologically active compounds, useful for diverse therapeutic areas.¹ The 1,2,3-triazole motif is associated with diverse pharmacological activities, such as antibacterial, antifungal, hypoglycemic, antihypertensive and analgesic properties. Polysubstituted five-membered aza heterocycles rank as the most potent glycosidase inhibitors.² Further, this nucleus in combination with or in linking with various other classes of compounds such as amino acids, steroids, aromatic compounds, carbohydrates etc., became prominent in having various pharmacological properties.³ 1,2,3-Triazole modified carbohydrates have became easily available after the discovery of the Cu(I)-catalyzed azide-alkyne 1,3-dipolar cycloaddition reaction⁴ and quickly became a prominent class of non-natural sugar derivatives.³ 1,2,3-Triazoles are one of the most important classes of heterocyclic organic compounds, which are reported to be present in a plethora of biologically active compounds, useful for diverse therapeutic areas.¹ The 1,2,3-triazole motif is associated with diverse pharmacological activities, such as antibacterial, antifungal, hypoglycemic, antihypertensive and analgesic properties. Polysubstituted five-membered aza heterocycles rank as the most potent glycosidase inhibitors.² Further, this nucleus in combination with or in linking with various other classes of compounds such as amino acids, steroids, aromatic compounds, carbohydrates etc., became prominent in having various pharmacological properties.³ 1,2,3-Triazole modified carbohydrates have became easily available after the discovery of the Cu(I)-catalyzed azide-alkyne 1,3-dipolar cycloaddition reaction⁴ and quickly became a prominent class of non-natural sugar derivatives.³

Thiazolidinones and 1,2,3-triazoles represent important classes of drugs in medicinal chemistry. They are among the most extensively investigated compounds by biochemists and medicinal chemists.⁶ Thiazolidinones in particular show interesting anticancer,⁷ anti-HIV,⁸ tuberculostatic,⁹ antihistaminic,¹⁰ anticonvulsant,¹¹ antibacterial,¹² and anti arrhythmic¹³ activities.

So called hybrid molecules have been shown to be highly active and effective in medicinal chemistry. Synergistic effects are obtained via hybridization of two different bioactive moieties with complementary pharmacophoric functions, or with different modes of action.¹⁴ The confirmation of this hypothesis has been well established in previous studies of 4-thiazolidinones coupled with other heterocyclic fragments,¹⁵ i.e. resulting in high antitumor activity.

Microwave irradiation is an alternative heating technique based on the transformation of electromagnetic energy into heat. Often this method increases the rate of chemical reactions¹⁶ and results in higher yields. In recent years, multi component reactions (MCRs)¹⁷ have received interesting attention due to their simplicity, efficiency, atom economy, shortened reaction times, and the possibility for diversity oriented synthesis.
Following the successful introduction of triazoles and thiazolidinones, microwave-assisted MCR reactions, inspired by the biological profile of triazoles, thiazolidinones, and in the continuation of our work on biologically active heterocycles\textsuperscript{18–29} we have developed a series of novel triazole-linked thiazolidenone derivatives, we have investigated the application of microwave irradiation for the synthesis of our hybrid molecules and evaluated their anticancer activity.

### 2. Result and Discussion

Di-acetone D-mannose (1), prepared from D-(+)-mannose by treating with acetone in the presence of a catalytic amount of sulphuric acid according to the literature procedure,\textsuperscript{30} on subsequent propargylation in DMF in the presence of NaH in 1 h afforded propargyl ether 2 (80%). Next, the propargyl ether was converted into triazole 3.
3. Cytotoxicity Evaluation Against Different Cancer Cell Lines

The cytotoxic effect of the compounds was tested by performing a Sulforhodamine B Assay (SRB) on different representative cell lines. Initially, the cell line of interest was seeded in a flat bottom 96-well plate (5000 cells/100 μL) in a medium containing 10% serum, followed by incubation for 18–20 h in an incubator continuously supplied with 5% CO₂, so as to ensure proper adherence of the cells to the surface bottom of the wells. After 18 h the cells were treated with the compound. Working dilutions of concentration of the compounds were prepared, of which 2 μL aliquot was added to each well, thereby making the final concentration of compound 0 to 100 μM. Each compound was tested in triplicate and the cytotoxicity was determined as the average of that triplicate. DMSO and doxorubicin (as standard control anti cancer drug) were taken as vehicle and positive controls, respectively. Further, the plates were incubated for another 48 h in an incubator maintained at 37 °C with a constant supply of 5% CO₂. After the period of 48 h, the cells were fixed using 10% TCA solution and incubated for 1 h at 4 °C after which the plate was rinsed carefully with MQ water and air dried; this was followed by addition of 0.057% SRB solution which was kept for approx. 30 min before it was rinsed off using 1% acetic acid. The plates were then air dried and the absorbance was measured using Perkin–Elmer Multimode Reader at 510 nm. To measure the absorbance, 100 μL of 10 mM Tris Base was added to each well to solubilize the SRB. The value of absorbance is directly proportional to cell growth and is thus used to calculate the IC₅₀ values. In this study for initial screening, four types of cancer cell lines, i.e. human lung cancer (A549), human breast cancer (MCF-7), prostate cancer (DU145) and HeLa cell lines were tested for the cytotoxic effect of the series of compounds. Based on the IC₅₀ values obtained, the compound 7b was picked for further assays to ascertain its effect on prostate cancer cell line (DU145).

3.1. Change in Morphology

Based on the cytotoxic ability of the compound, its effect on the morphology of the cells was also ascertained. To achieve this, a 24-well plate was seeded with cells in a manner previously described and incubated for 18–20 h. Then, the cells were treated with increasing concentra-
of 7b. After another 48 h of incubation, the experiment was terminated and the cells were observed under the microscope and images were captured using Olympus Xi71 microscope.

3.2. Colony Formation Assay

The long term effect of the 7b on the anchorage independent nature of cancer cells was further tested in the following experiment. The experiment was a soft agar assay which was conducted as reported previously with minor modifications. In the experiment, base agar was prepared by mixing 1% of agarose (Bacto Agar: Becton, Dickinson, Sparks, MD) with 2 x DMEM along with 20% FBS and 2X antibiotics in 6-well plates in order to achieve final concentration of 0.5% of agar in 1X growth medium with 10% serum concentration. After the solidification of the base agar, 2.5 x 10^4 cells were mixed with cultivation medium containing compound at varying concentrations along with agar solution to obtain a final concentration of 0.35% agar. This was spread on top of the base agar previously solidified. The plate was incubated for 9 days with periodic replenishment every 3 days with medium and compound. Over the period of time, plates were monitored regularly for appearance of colonies. After 9 days of incubation the plates were stained with 0.005% crystal violet solution until colonies turned purple in color. The excess stain was washed off using MQ water and the colonies were photographed and counted using a microscope.

3.3. Determination of Caspase-3 and Caspase-9 Activities

Caspase activity, specifically, caspase-9 and caspase-3 activities were analyzed in the cell lysates obtained from DU 145 cells previously treated with compound 7b. The activity was observed using fluorogenic substrates, namely Ac-DEVD-AMC and Ac-LEHD-AFC for caspase-3 and caspase-9, respectively. After 48 h treatment of cells with compound 7b, harvested cells were lysed directly in caspase lyses buffer (50 mM HEPES, 5 mM CHAPS, 5 mM DTT, pH 7.5). The lysates were incubated with the respective substrate (Ac-DEVD-AFC/Ac-LEHD-AMC) in 20 mM HEPES (pH 7.5), 0.1% CHAPS, 2 mM EDTA and 5 mM DTT at 37 °C for 2 h. The release of AFC and AMC was analyzed by a fluorimeter using an excitation/emission wavelength of 400/505 nm (for AFC) and 380/460 nm (for AMC) which is directly proportional to caspase-9 and caspase-3 activity, respectively. The observed fluorescence values were normalized with total protein concentration estimated by Bradford method and the relative caspase activities were calculated as the ratio of values between mock treated (DMSO) and treated cells.

3.4. Senescence Assay

Compounds with anti-cancer potential may have the possibility to induce senescence in cells, thus limiting their proliferation. The ability to induce cellular senescence was determined by measuring senescence-associated beta-galactosidase (SA-βgal) activity (pH 6.0) in DU145 cells exposed to compound 7b. A 24-well plate was seeded with DU 145 cells as previously described and treated with the compound and subsequently incubated for 48 h. After 48 h, the media was aspirated and the cells were washed with PBS (2 x 1 min) and fixed by adding enough volume of fixation solution (2% formaldehyde and 0.2% glutaraldehyde in PBS solution) to submerge the cells in solution. After incubation for 5 min at room temperature
the fixation solution was removed and the cells were washed twice with PBS (2 × 1 min). The resultant fixed cells were then stained with freshly prepared staining solution (40 mM citric acid/Na phosphate buffer, 5 mM K₄[Fe(CN)₆]·3H₂O, 5 mM K₃[Fe(CN)₆], 150 mM sodium chloride, 2 mM magnesium chloride and 1 mg X-gal in 1 mL distilled water) overnight at 37 °C. The excess stain was removed by repeated washings with PBS and plate was allowed to dry at room temperature. The cells stained with SA-βgal levels were observed and photographed under an Olympus Xi71 microscope.

3. 5. PI Uptake for Analysis of Cell Death

Cell death induced by compound 7b was determined as a measure of PI uptake. Cells were harvested after treatment with compound at desired concentration and fixed in 70% ethanol at -20 °C overnight. The cells were then collected in the form of pellet. All cells in the form of a pellet were then resuspended in PI solution (RNase 0.1 mg/mL, Triton X-100 0.05%, PI 50 μg/mL) and incubated for 1 h in dark at room temperature. The excess PI solution was washed away by repeated washings with PBS buffer. The resultant PI uptake was analyzed by fluorescence activated cell sorting (FACS Caliber System; BD Bio-science, Erembodegem, Belgium) in a FL-2 fluorescence detector (10000 events were recorded

<table>
<thead>
<tr>
<th>Compound</th>
<th>G0/G1</th>
<th>S</th>
<th>G2/M</th>
</tr>
</thead>
<tbody>
<tr>
<td>DMSO</td>
<td>71.36</td>
<td>5.02</td>
<td>22.74</td>
</tr>
<tr>
<td>5 μM</td>
<td>63.92</td>
<td>8.10</td>
<td>22.16</td>
</tr>
<tr>
<td>10 μM</td>
<td>69.15</td>
<td>5.54</td>
<td>24.65</td>
</tr>
<tr>
<td>15 μM</td>
<td>71.86</td>
<td>3.92</td>
<td>22.51</td>
</tr>
<tr>
<td>20 μM</td>
<td>72.11</td>
<td>3.89</td>
<td>20.05</td>
</tr>
<tr>
<td>25 μM</td>
<td>76.48</td>
<td>2.99</td>
<td>18.77</td>
</tr>
</tbody>
</table>

Table 3. Compound 7b induced G0/G1 phase cell cycle arrest in DU145 cells. Cells were treated with varying concentrations of compound 7b (5, 10, 15, 20 and 25 μM) for 48 h and cell cycle progression was examined by flow cytometry. Table shows the percentage cell fractions in G0/G1, S and G2/M phases of compound 7b treated DU145 cells.

Figure 1. DU145 cell were treated with compound 7b at indicated concentration or DMSO. Upon exposure of DU145 cells to compound 7b the extent of change in cell morphology of cells is observed with increasing concentration.
Flow cytometry data was analyzed using FCS express 4 software (De Novo Software, Los Angeles, CA).

4. Experimental

Commercial grade reagents were used as supplied. Solvents, except those of analytical grade, were dried and purified according to literature when necessary. Reaction progress and purity of the compounds were checked by thin-layer chromatography (TLC) on pre-coated silica gel F254 plates from Merck and compounds visualized either by exposure to UV light or dipping in 1% aqueous potassium permanganate solution. Silica gel chromatographic columns (60–120 mesh) were used for separations. Microwave reactions were carried out in mini lab microwave catalytic reactor (ZZKD, WBFY-201) and reaction mixture temperatures were measured through an immersed fibre-optic sensor. All melting points are uncorrected and measured using Fisher–Johns apparatus. IR spectra were recorded as KBr disks on a Perkin–Elmer FT IR spectrometer. The $^1$H NMR and $^{13}$C NMR spectra were recorded on a Varian Gemini spectrometer (300 MHz for $^1$H and 75 MHz for $^{13}$C). Chemical shifts are reported as $\delta$ ppm against TMS as internal reference and coupling constants ($J$) are reported in Hz units. Mass spectra were recorded on a VG micro mass 7070H spectrometer. Elemental analyses (C, H, N) determined by a Perkin–Elmer 240 CHN elemental analyzer, were within $\pm0.4\%$ of theoretical.
(3aS,4R,6S,6aS)-4-\((R)-2,2\text{-dimethyl-1,3-dioxolan-4-yl})-2,2\text{-dimethyl-6-\((prop-2-ynyloxy)tetrahydrofu-
ro[3,4-d][1,3]dioxole (2). Sodium hydride (60% in miner-
al oil, 0.64 g) was added to a stirred solution of 3 (3.6 g, 13.84 mmol) in DMF (80 mL) at 0 °C and allowed to stir
for 30 min. This yellow mixture was cooled to 0°C and treated with propargyl bromide (4.2 g) in DMF (20 mL). The
dark brown reaction mixture was allowed to stir for an hour at room temperature and quenched (at 5–10 °C) with saturated aqueous ammonium chloride (20 mL). The

Figure 4. Senescence induced by compound 7b was quantified using SA-βgal-staining. As shown in the figure, 7b did not induce senescence in cells as at higher concentrations the cells underwent apoptosis.

Figure 5. Cell cycle analysis of DU145 cells treated with compound 7b. Cells were treated with either DMSO or 7b and the DNA content was measured by propidium iodide staining to determine the distribution of cells in various phases of cell cycle. DMSO was taken as reference.
crude product was extracted with CH₂Cl₂ (3 × 30 mL), dried (Na₂SO₄) and concentrated. The residue was purified by column chromatography on silica gel (5% EtOAc : hexane) to afford 4 (3.1 g, 75%) as viscous oil. IR (KBr): ν 3312, 3297, 2992, 2965, 2936, 2922, 1630, 1544, 1510, 1212, 1161, 1022, 732 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 8.01 (s, 1H, Ar-H), 7.51 (d, J = 9.2 Hz, 2H, Ar-H), 7.40 (d, J = 8.9 Hz, 2H, Ar-H), 5.49 (d, J = 3.7 Hz, 1H, C₆H), 4.52 (t, J = 3.9 Hz, 1H, C₆H), 4.58 (s, 2H, OCH₂), 3.88–3.81 (m, 2H, C₆H, C₆H), 4.01–3.92 (m, 3H, 3H, 2 × C₂H), 2.42 (bs, 1H, OCH₂), 1.50 (s, 3H, CH₃), 1.45 (bs, 1H, OH), 1.34 (s, 3H, CH₃); ¹³C NMR (75 MHz, CDCl₃): δ 143.2, 133.2, 122.1, 117.2, 110.2, 109.2, 102.1, 78.8, 77.1, 75.1, 70.6, 67.2, 65.2, 63.2, 26.6, 26.24. MS: m/z (M⁺H) 412. Anal. Calcd for C₁₃H₁₂ClCIN₂O₇: C, 52.49; H, 5.38; N, 10.21. Found: C, 52.35; H, 5.25; N, 10.211.

2-((3aR,4S,6S,6aS)-6-((1-(4-Chlorophenyl)-1H-1,2,3-triazol-4-yl)methoxy)-2,2-dimethyltetrahydrofurano[3,4-d][1,3]dioxol-4-yl)-3-phenylthiazolidin-4-one 6a–g. To a solution of diol 4 (0.200 g, 0.48 mmol) in CH₂Cl₂ (5 mL), NaO₂ (0.130 g, 0.61 mmol) was added at 0 °C and stirred at room temperature for 6 h. The reaction mixture was filtered and washed with CH₂Cl₂ (2 × 10 mL). It was dried (Na₂SO₄) and evaporated to give aldehyde 5 (0.150 g) in quantitative yield as a yellow liquid, which was used as such for the next reaction.

To a stirred mixture of 5 (0.150 g, 0.395 mmol), aromatic amine (0.395 mmol) and anhydrous thioglycolic acid (0.160 g, 0.211 mmol) in dry toluene (5 mL), ZnCl₂ (0.100 g, 0.751 mmol) was added after 2 min and irradiated in microwave bath reactor at 280 W for 4–7 minutes at 110 °C. After cooling, the filtrate was concentrated to dryness at reduced pressure and the residue was taken up in ethyl acetate. The ethyl acetate layer was washed with 5% sodium bicarbonate solution and finally with brine. The organic layer was dried over Na₂SO₄ and evaporated to dryness at reduced pressure. The crude product thus obtained was purified by column chromatography on silica gel (60–120 mesh) with hexane – ethyl acetate as eluent. Under conventional method the reaction mixture in toluene (10 mL) was refluxed at 110 °C for the appropriate time (Table 1).

(R)-1-((3aS,4R,6S,6aS)-6-((1-(4-Chlorophenyl)-1H-1,2,3-triazol-4-yl)methoxy)-2,2-dimethyltetrahydrofurano[3,4-d][1,3]dioxol-4-yl)-ethane-1,2-diol (4). A mixture of 3 (3 g, 6.65 mmol) in 60% sq. AcOH (25 mL) was stirred at room temperature for 12 h. Reaction mixture was neutralized with anhy. NaHCO₃ (15 g) and extracted with EtOAc (3 × 41 mL). The combined organic layers were dried (Na₂SO₄), evaporated and residue purified by column chromatography (60–120 mesh silica gel, 41% ethyl acetate in pet. ether) to afford 4 (2.6 g, 82%) as a pale yellow solid; mp 168–171 °C. IR (KBr) ν 3218, 3486, 3362, 3292, 2965, 2936, 2922, 1630, 1544, 1510, 1212, 1161, 1022, 732 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 8.01 (s, 1H, Ar-H), 7.51 (d, J = 9.2 Hz, 2H, Ar-H), 7.40 (d, J = 8.9 Hz, 2H, Ar-H), 5.49 (d, J = 3.7 Hz, 1H, C₆H), 4.52 (t, J = 3.9 Hz, 1H, C₆H), 4.58 (s, 2H, OCH₂), 3.88–3.81 (m, 2H, C₆H, C₆H), 4.01–3.92 (m, 3H, 3H, 2 × C₂H), 2.42 (bs, 1H, OCH₂), 1.50 (s, 3H, CH₃), 1.45 (bs, 1H, OH), 1.34 (s, 3H, CH₃); ¹³C NMR (75 MHz, CDCl₃): δ 143.2, 133.2, 122.1, 117.2, 110.2, 109.2, 102.1, 78.8, 77.1, 75.1, 70.6, 67.2, 65.2, 63.2, 26.6, 26.2. MS: m/z (M⁺H) 412. Anal. Calcd for C₁₃H₁₂ClCIN₂O₇: C, 52.49; H, 5.38; N, 10.21. Found: C, 52.35; H, 5.25; N, 10.211.
methyltetrahydrofuro[3,4-d][1,3]dioxol-4-yl]thiazolidin-4-one (6b), mp 206–208 °C; IR (KBr) v 3430, 3219, 2974, 2972, 2812, 1710, 1610, 1546, 1510, 1409, 1219, 682 cm⁻¹; 1H NMR (300 MHz, CDCl₃): δ 8.01 (s, 1H, Ar-H), 7.46 (d, J = 9.2 Hz, 4H, Ar-H), 7.41 (d, J = 8.9 Hz, 4H, Ar-H), 5.62 (d, J = 3.6 Hz, 1H, C-H), 4.84 (d, J = 5.2 Hz, CH-S), 4.50 (t, J = 3.9 Hz, 1H, C-H), 4.49 (s, 2H, OCH₂), 3.86–3.71 (m, 1H, C₄H), 3.66 (s, 2H, CH₂), 3.29 (dd, J = 9.1, 4.2 Hz, 1H, C₄H), 1.45 (s, 3H, CH₃), 1.32 (m, 3H, CH₃); 13C NMR (75 MHz, CDCl₃): 170.2, 138.4, 134.4, 133.0, 128.4, 127.6, 125.2, 122.1, 118.4, 111.6, 104.3, 80.5, 74.1, 65.3, 52.1, 34.3, 25.5; MS: m/z (M⁺H) 563. Anal. Caled for C₂₅H₂₄Cl₂N₄O₅S: C, 53.29; H, 4.29; N, 9.49. Found: C, 55.21; H, 4.16; N, 9.83.

2-((3aR,4S,6S,6aS)-6-((1-(4-Chlorophenyl)-1H-1,2,3-triazol-4-yl)methoxy)-2,2-dimethyltetrahydrofuro[3,4-d][1,3]dioxol-4-yl)-3-(4-nitrophenyl)thiazolidin-4-one (6c), mp 201–205 °C; IR (KBr) v 3432, 3216, 2984, 2961, 2820, 1710, 1605, 1536, 1512, 1416, 1372, 1271, 863, 630 cm⁻¹; 1H NMR (300 MHz, CDCl₃): 8.16 (d, J = 8.7 Hz, 2H), 8.02 (s, 1H, Ar-H), 7.49 (d, J = 9.2 Hz, 2H, Ar-H), 7.40 (d, J = 8.5 Hz, 2H, Ar-H), 6.72 (d, J = 9.8 Hz, 2H, Ar-H), 5.69 (d, J = 3.6 Hz, 1H, C-H), 4.86 (d, J = 5.2 Hz, CH-S), 4.52 (t, J = 3.9 Hz, 1H, C-H), 4.49 (s, 2H, OCH₂), 3.86–3.81 (m, 1H, C₄H), 3.66 (s, 2H, CH₂), 3.24 (dd, J = 9.1, 4.2 Hz, 1H, C₄H), 1.50 (s, 3H, CH₃), 1.24 (m, 3H, CH₃); 13C NMR (75 MHz, CDCl₃): δ 170.2, 145.5, 145.5, 142.2, 134.2, 133.0, 124.6, 124.3, 121.4, 118.8, 111.4, 104.6, 80.5, 77.2, 73.8, 66.4, 52.1, 34.2, 26.2; MS: m/z (M⁺H) 574. Anal. Caled for C₂₆H₂₇ClN₄O₅S: C, 53.21; H, 4.21; N, 12.20. Found: C, 53.26; H, 4.19; N, 12.11.

2-((3aR,4S,6S,6aS)-6-((1-(4-Chlorophenyl)-1H-1,2,3-triazol-4-yl)methoxy)-2,2-dimethyltetrahydrofuro[3,4-d][1,3]dioxol-4-yl)-3-(4-nitrophenyl)thiazolidin-4-one (6d), mp 181–183 °C; IR (KBr) v 3436, 3234, 2986, 2976, 2834, 1710, 1705, 1610, 1549, 1516, 1418, 1262, 865 cm⁻¹; 1H NMR (300 MHz, CDCl₃): δ 8.13 (d, J = 8.7 Hz, 2H, Ar-H), 8.01 (s, 1H, Ar-H), 7.50 (d, J = 9.2 Hz, 2H, Ar-H), 7.35–6.92 (m, 5H, Ar-H), 5.64 (d, J = 3.6 Hz, 1H, C-H), 4.84 (d, J = 5.2 Hz, 1H, CH-S), 4.52 (t, J = 3.9 Hz, 1H, C-H), 4.44 (s, 2H, OCH₂), 3.86–3.71 (m, 1H, C₄H), 3.66 (s, 2H, CH₂), 3.16 (dd, J = 9.1, 4.2 Hz, 1H, C₄H), 2.08 (s, 3H, CH₃), 1.49 (s, 3H, CH₃), 1.26 (m, 3H, CH₃); 13C NMR (75 MHz, CDCl₃): δ 170.2, 143.6, 136.7, 134.3, 133.3, 130.2, 129.1, 127.6, 125.2, 122.1, 118.8, 111.2, 104.4, 81.4, 78.3, 74.4, 66.1, 52.1, 26.2, 16.3; MS: m/z (M⁺H) 545. Anal. Caled for C₂₆H₂₅ClN₄O₅S: C, 57.51; H, 5.51; N, 10.32. Found: C, 56.86; H, 5.39; N, 10.11.

2-((3aR,4S,6S,6aS)-6-((1-(4-Chlorophenyl)-1H-1,2,3-triazol-4-yl)methoxy)-2,2-dimethyltetrahydrofuro[3,4-d][1,3]dioxol-4-yl)-3-(4-nitrophenyl)thiazolidin-4-one (6e), mp 181–183 °C; IR (KBr) v 3418, 3202, 2981, 2972, 2830, 1702, 1691, 1610, 1536, 1519, 1412, 1251, 856 cm⁻¹; 1H NMR (300 MHz, CDCl₃): δ 8.2 (d, J = 8.7 Hz, 2H, Ar-H), 8.01 (s, 1H, Ar-H), 7.44 (d, J = 9.2 Hz, 2H, Ar-H), 7.36 (d, J = 8.3 Hz, 2H, Ar-H), 7.16 (d, J = 8.3 Hz, 2H, Ar-H), 5.66 (d, J = 3.6 Hz, 1H, C-H), 4.86 (d, J = 5.2 Hz, 1H, CH-S), 4.56 (t, J = 3.9 Hz, 1H, C-H), 4.54 (s, 2H, OCH₂), 3.86–3.81 (m, 1H, C₄H), 3.66 (s, 2H, CH₂), 3.16 (dd, J = 9.1, 4.2 Hz, 1H, C-H), 2.32 (s, 3H, CH₃), 1.43 (s, 3H, CH₃), 1.36 (m, 3H, CH₃); 13C NMR (75 MHz, CDCl₃): δ 171.6, 143.1, 136.4, 131.6, 130.3, 129.2, 128.2, 127.9, 124.2, 121.1, 118.2, 101.2, 80.9, 78.3, 74.6, 64.9, 51.6, 26.4, 15.1; MS: m/z (M⁺Na) 565. Anal. Caled for C₂₇H₂₆ClN₅O₅S: C, 57.51; H, 5.51; N, 10.32. Found: C, 56.82; H, 5.35; N, 10.09.

Srinivas et al.: Microwave-assisted Synthesis of Hybrid Heterocycles ...
mmol) in CH₂Cl₂ (5 mL), NaIO₄ (0.130 g, 0.61 mmol) was added at 0 °C and stirred at room temperature for 6 h. The reaction mixture was filtered and washed with CH₂Cl₂ (2 × 10 mL). It was dried (Na₂SO₄) and evaporated to give aldehyde 7 (0.150 g) in quantitative yield as a yellow liquid, which was used as such for the next reaction.

To a stirred mixture of 7 (0.150 g, 0.395 mmol), aromatic amine (0.395 mmol) and thiolic acid (0.125 g, 0.86 mmol) in dry toluene (5 mL), anhydrous ZnCl₂ (0.100 g, 0.751 mmol) was added after 2 min and irradiated in microwave bath reactor at 280 W for 4–7 minutes at 110 °C. After cooling, the filtrate was concentrated to dryness under reduced pressure and the residue was taken up in ethyl acetate. The ethyl acetate layer was washed with saturated Na₂CO₃ solution and finally with brine. The organic layer was dried over Na₂SO₄ and evaporated to dryness under reduced pressure. The crude product thus obtained was purified by column chromatography on silica gel (60–120 mesh) with hexane - ethyl acetate as eluent.

Under conventional method the reaction mixture in toluene (10 mL) was refluxed at 110°C for the appropriate time (Table 1).

2-(2-((3a,4R,6S,6aS)-6-((1-(4-Chlorophenyl)-1H-1,2,3-triazol-4-yl) methoxy)-2,2-dimethyltetrahydrofuro[3,4-d][1,3]dioxol-4-yl)-4-oxothiazolidin-5-yl)acetic acid (7a). IR (KBr) ν 3434, 3421, 3216, 2972, 2984, 2970, 2822, 2822, 1722, 1716, 1610, 1539, 1512, 1410, 1214, 861, 684 cm⁻¹; 1H NMR (300 MHz, CDCl₃): δ 11.34 (s, 1H, CO₂H), 8.05 (s, 1H, Ar-H), 7.45 (d, J = 9.2 Hz, 2H, Ar-H), 7.38 (d, J = 8.9 Hz, 2H, Ar-H), 7.32–7.28 (m, 5H, Ar-H), 6.05 (s, 1H, CHS), 5.53 (d, J = 4.2 Hz, 1H, H), 4.59 (t, J = 3.9 Hz, 1H, C₂H), 4.55 (t, J = 4.2 Hz, 1H, C₂H), 4.42 (s, 2H, OCH₂), 3.82–3.76 (m, 1H, C₄H), 3.12 (dd, J = 3.9 Hz, 1H, C₂H), 2.24 (d, 2H, CH₂), 1.43 (s, 3H, CH₃), 1.20 (m, 3H, CH₃); 13C NMR (75 MHz, CDCl₃): δ 170.3, 143.2, 141.9, 132.9, 124.4, 122.2, 118.2, 110.3, 103.2, 80.1, 76.9, 72.8, 65.1, 50.0, 36.2, 32.9, 24.9; MS: m/z (M⁺+H) 545. Anal. Calcd for C₂₇H₂₅Cl₂N₄O₇S: C, 55.24; H, 4.64; N, 9.54. Found: C, 55.12; H, 4.59; N, 9.39.

2-(2-(3-(4-Chlorophenyl)-2-(2-((3a,4R,6S,6aS)-6-((1-(4-Chlorophenyl)-1H-1,2,3-triazol-4-yl) methoxy)-2,2-dimethyltetrahydrofuro[3,4-d][1,3]dioxol-4-yl)-4-oxothiazolidin-5-yl)acetic acid (7a)). mp 221–241 °C; IR (KBr) ν 3428, 3421, 3216, 2972, 2819, 1721, 1715, 1606, 1529, 1509, 1410, 1206, 679 cm⁻¹; 1H NMR (300 MHz, CDCl₃): δ 11.24 (s, 1H, CO₂H), 7.88 (s, 1H, Ar-H), 7.35 (d, J = 9.2 Hz, 4H, Ar-H), 7.39 (d, J = 8.9 Hz, 4H, Ar-H), 6.10 (s, 1H, CHS), 5.78 (d, J = 4.2 Hz, 1H, C₂H), 4.72 (t, J = 3.9 Hz, 1H, C₂H), 4.55 (t, 1H, CH), 4.52 (s, 2H, OCH₂), 3.92–3.89 (m, 1H, CH₃), 3.20 (dd, J = 9.1, 4.2 Hz, 1H, CH), 2.34 (d, 2H, CH₂), 1.53 (s, 3H, CH₃), 1.30 (m, 3H, CH₃); 13C NMR (75 MHz, CDCl₃): δ 170.4, 142.2, 141.6, 132.6, 127.8, 125.9, 121.2, 117.4, 103.5, 80.4, 77.23, 71.2, 66.1, 51.3, 35.9, 31.2, 24.6; MS: m/z (M⁺+H) 621. Anal. Calcd for C₁₉H₁₅Cl₂N₄O₇S: C, 52.18; H, 4.22; N, 9.01. Found: C, 52.02; H, 4.09; N, 8.95.

Srinivas et al.: Microwave-assisted Synthesis of Hybrid Heterocycles ...
C(H), 2.32 (d, 2H, CH₂), 2.16 (s, 3H, CH₃), 1.47 (s, 3H, CH₂); ¹³C NMR (75 MHz, CDCl₃): 190.4, 172.4, 143.6, 141.9, 132.9, 124.4, 121.2, 117.2, 103.6, 80.6, 76.4, 72.4, 65.5, 52.1, 35.3, 32.2, 24.2, 15.2; MS: m/z (M⁺+H) 600. Anal. Calcd for C₃₈H₅₈ClN₄O₈S: C, 53.78; H, 4.52; N, 9.29. Found: C, 53.52; H, 4.35; N, 8.99.

5. Conclusion

A series of novel triazole linked thiazolidenone derivatives 6a–g and 7a–g was prepared and evaluated for their anticancer activity. The screened compound 7b exhibited potent anticancer activity compared to standard drug at the tested concentrations.

6. Acknowledgements

The authors are thankful to CSIR-New Delhi for the financial support (Project funding No: 02(247)15/EMR-II). Director, CSIR- IICT, Hyderabad, India, for NMR and MS spectral analysis and Principal Vaagdevi Degree and PG College for his constant encouragement to carry out research work.

7. References

(d) W. G. Lewis, G. Green, F. Z. Grynszpan, Z. Radić, P. R. Carlier, P. Taylor, M. G. Finn, K. B. Sharpless, Angew.

---

Srinivas et al.: Microwave-assisted Synthesis of Hybrid Heterocycles
S postopkom sinteze v eni sami posodi smo s pomočjo kondenzacije (3aS,4S,6S,6aS)-6-{[(1-(4-klorofenil)-1H-1,2,3-triazol-4-il)metoxi]-2,2-dimetiltetrahidrofuro[3,4-d][1,3]dioksol-4-karbaldehida 5 z merkapto kislinami in primarnimi amini v prisotnosti ZnCl₂ pripravili serijo novih hibridnih heterociklov 6a–g in 7a–g. Sinteze so bile izvedene tako pod mikrovalovnimi kot tudi konvencionalnimi pogoji segrevanja. Spojino 5 smo pripravili iz di-aceton D-manoze s pomočjo »click« reakcije, s sledečo odstranitvijo primarne acetonidne zaščite in z oksidativnim razcepom. Karakterizacijo novih spojbin smo izvedli s pomočjo IR, NMR, MS in elementne analize. Za nove spojine smo določili tudi delovanje proti različnim rakastim celicam.